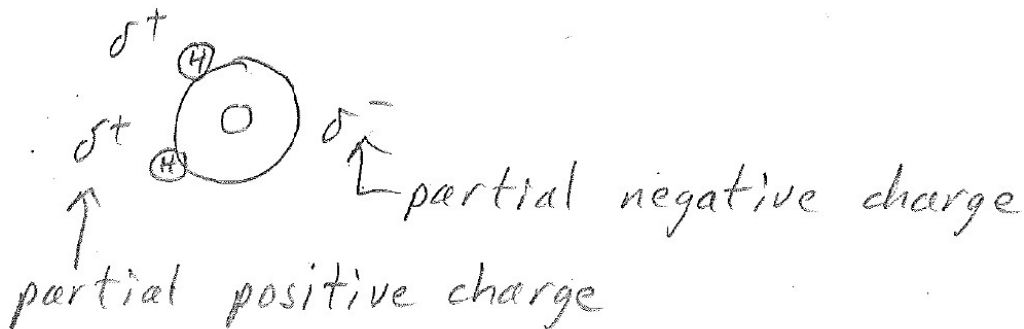


Resistive Sensing of Aqueous Solutions

1. Background

Water: H_2O → a molecule with a polar covalent bond

Physically:



Therefore, as a polar compound, H_2O has a permanent dipole, which causes ionic compounds (salts) to dissolve in it.

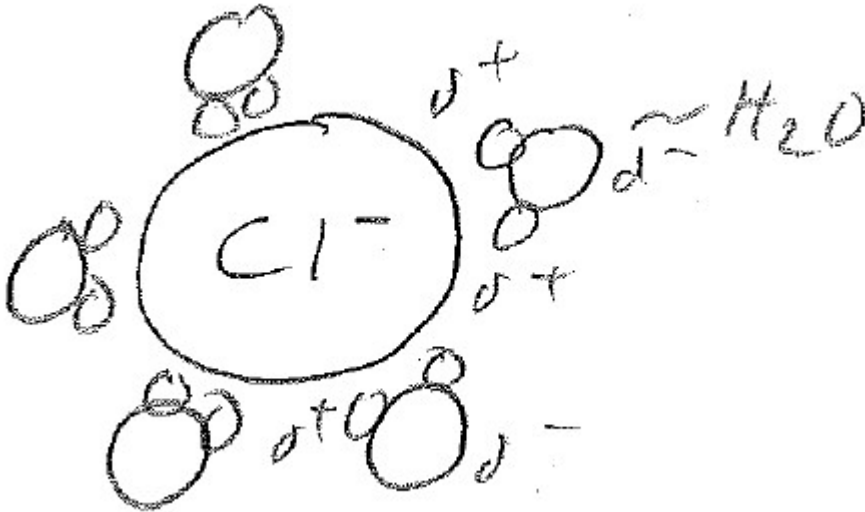
Example: table salt: $\text{NaCl} \rightarrow \text{Na}^+ + \text{Cl}^-$

Na^+ is a cation and Cl^- is an anion

In water: $\text{NaCl} + \text{H}_2\text{O} \rightarrow \text{Na}^+(\text{aq}) + \text{Cl}^-(\text{aq})$

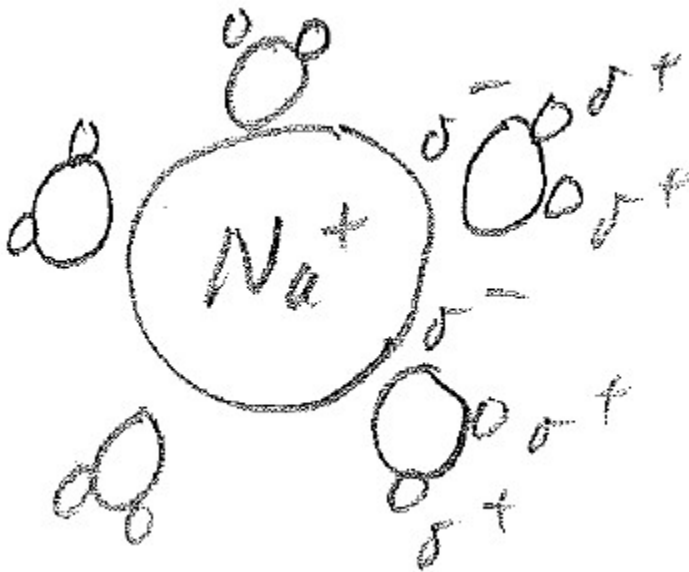
$\text{Na}^+(\text{aq})$ and $\text{Cl}^-(\text{aq})$ are hydrated ions

$\text{Cl}^-(\text{aq}) \rightarrow$ hydrated anion:



Observe that the partial positive charge on the H_2O molecules is drawn to the Cl^- anion in the exaggerated drawing. The resulting composite structure is called a hydration shell or a hydration sphere.

Similarly, the partial negative charge on the H_2O molecules is drawn to the Na^+ cation in the exaggerated drawing below, also forming a hydration shell. $\text{Na}^+(\text{aq}) \rightarrow$ hydrated cation:



Note: atom sizes are NOT to scale

2. Applying an electric field across the solution

$\text{Na}^+(\text{aq})$ and $\text{Cl}^-(\text{aq})$ act as mobile charge carriers, like electrons and holes in a semiconductor.

However, ion charge carriers are slow compared to electrons and holes. Therefore, an aqueous solution is a much lower frequency electrical system.

The higher the NaCl concentration, the higher the number of mobile charge carriers, and the higher the electrical conductivity.

The symbol for electrical conductivity is σ , the same symbol used for tensile stress in mechanical systems.

For low salt concentrations:

$\sigma \propto \text{ion concentration}$:

$$\sigma = \sum_{i=1}^n \lambda_i C_i$$

where: σ = solution electrical conductivity (EC)

λ_i = ionic conductivity of the i^{th} ion type

C_i = molar concentration of the i^{th} ion type

For high concentrations: ions will interact with each other, invalidating this equation:



By measuring EC, the concentration of dissolved salts can be determined.

3. Temperature effects

EC is affected by temperature nonlinearly. However, for $0^{\circ}\text{C} \leq T \leq 30^{\circ}\text{C}$, a linear model is:

$$\sigma_T = \sigma_{25}[1 + a(T - 25)]$$

Where: σ_{25} is the EC at 25°C

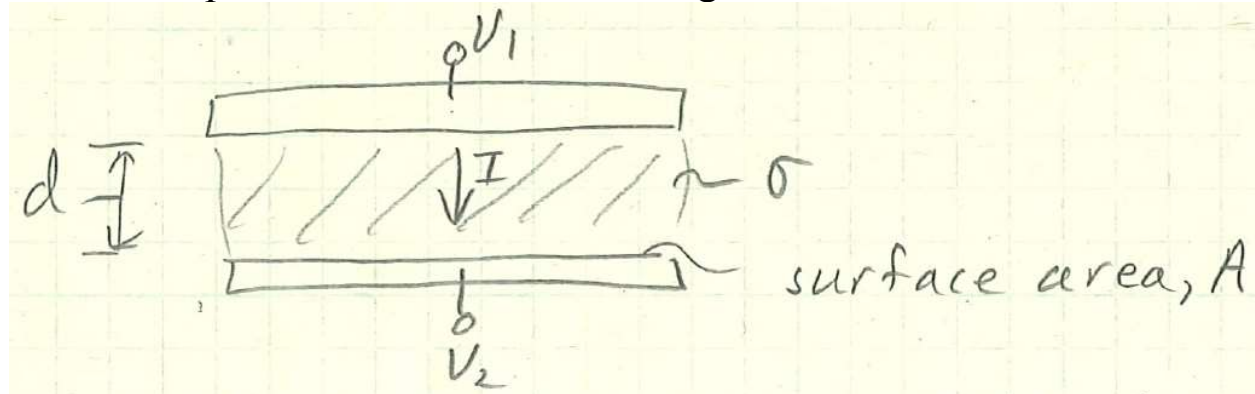
“a” is a temperature compensation factor, $[a] = 1/^{\circ}\text{C}$

Various values for “a” are found in publications: 0.0191, 0.02, 0.025, 0.0187, etc. However, all are approximately $a = 0.02$.

Often, σ_{25} is calculated and used for comparison.

4. Measuring EC of an aqueous solution

Consider 2 parallel electrodes in a homogeneous solution:



Assumptions: $A \gg d^2$, and a uniform current density

Therefore: $R = \frac{d}{\sigma A}$

Conductance: $G = \frac{1}{R} = \frac{\sigma A}{d} = \sigma \kappa$

The sensor's cell constant, κ , is a function of the sensor's geometry, and is independent of concentration or temperature.

$$[\kappa] = \text{m}$$

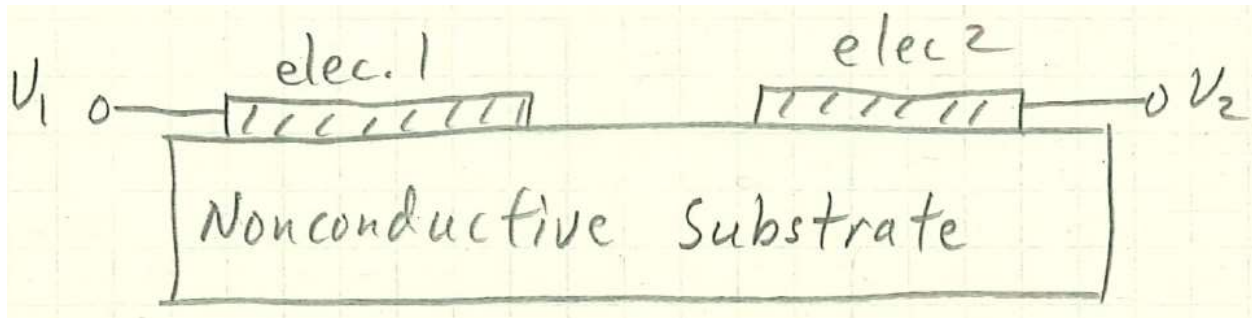
$$[G] = \text{S}$$

$$[\sigma] = \text{S/m}$$

If κ is known, you can measure G (or R) and calculate σ .

a. Flat electrode sensor

Consider:



This type of electrode structure is easy to build in MEMS or on a PCB.

However, it has a non-uniform current density. Therefore, κ is difficult to calculate.

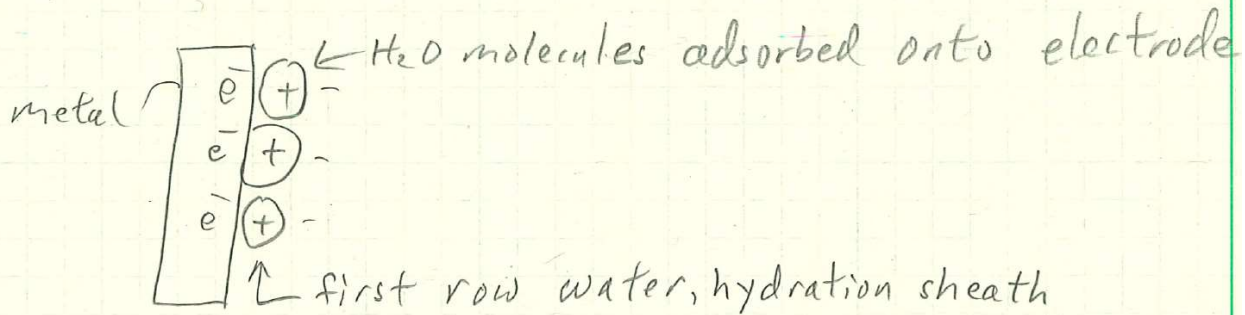
But, you can use an EC sensor of known κ and measure EC with it. Then measure G with the flat electrode sensor. Then calculate κ for the flat electrode sensor:

$$\kappa = \frac{G}{\sigma}$$

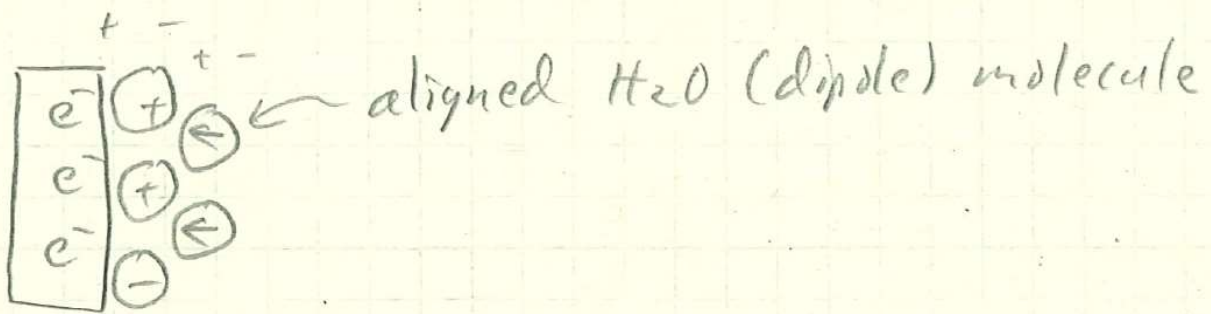
With now knowing κ for the flat electrode sensor, the sensor can be used to determine σ from any G measurement.

5. Frequency effects

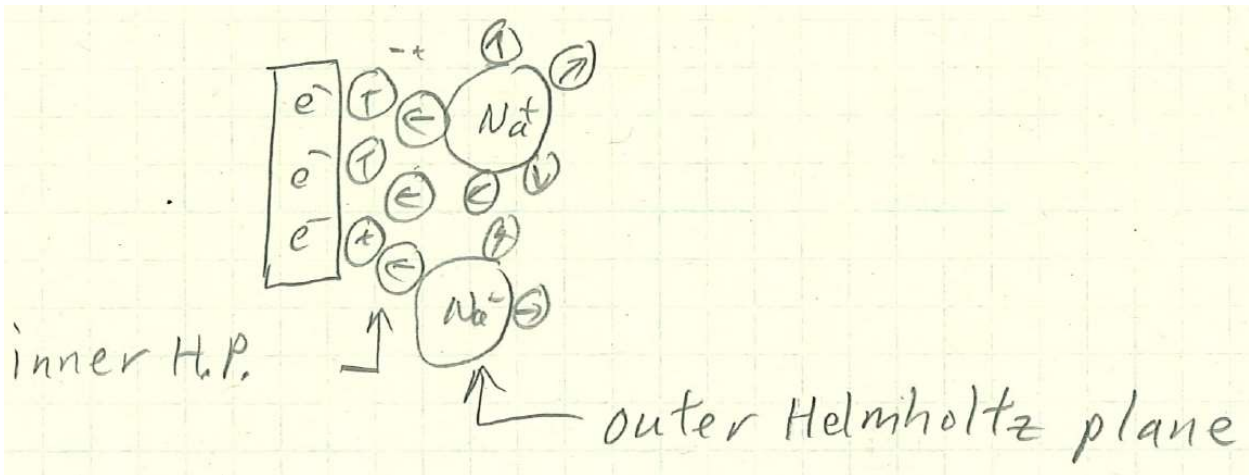
When a metal electrode is put in water, a water molecule making contact with the metal will transfer an electron to the metal, leaving the water molecule with a positive charge.



Other water molecules (dipoles) will be attracted to the H₂O molecules on the electrode, and will align with them.



Cations will be attracted to the aligned dipoles and will form hydrated cations there:

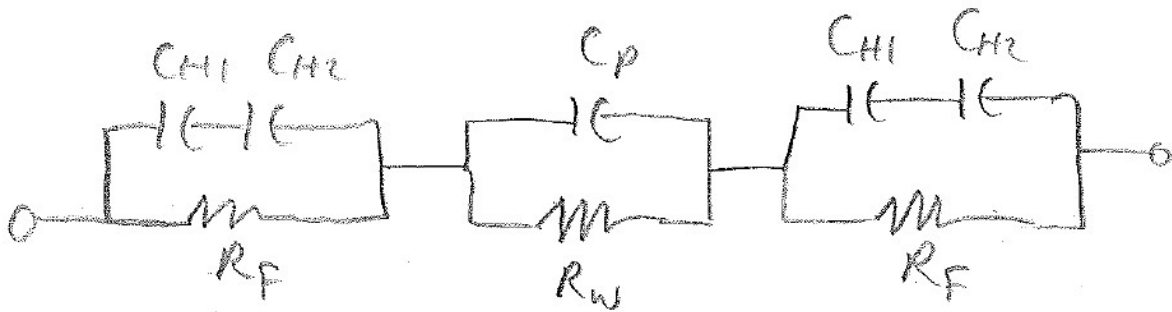


Some Na⁺ ions will also chemically bond to the metal, forming an inner Helmholtz plane.

The 2 Helmholtz planes together form the “electrical double layer,” or EDL, which is two capacitors in series (C_{H1} and C_{H2}), and in parallel with

the resistance (R_F) between the electrode and the water when a voltage is applied. The subcircuit consisting of C_{H1} , C_{H2} , and R_F is called the Faradaic impedance.

Circuit model:



$$\frac{1}{\frac{1}{C_{H1}} + \frac{1}{C_{H2}}} > C_p$$

The Faradaic impedance of the EDL results in an EC measurement error at DC and low frequencies.

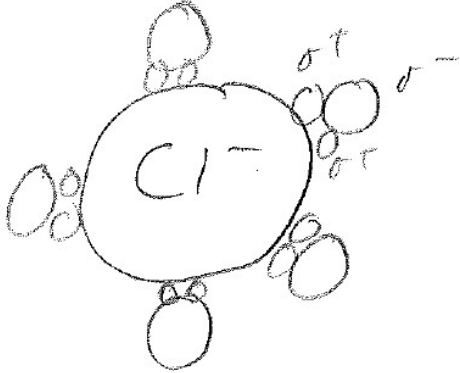
Measuring EC at a sufficiently high frequency shorts across the EDL for a more accurate EC reading.

But too high of a measurement frequency will result in shorting across the small parasitic capacitance, C_p .

Choose some measurement frequency that \sim shorts C_{H1} and C_{H2} , but not C_p . The goal is to measure R_w with the sensor. The exact frequency used depends on the sensor. With the sensors I have developed, 50 kHz to 1 MHz seems to work well.

6. More on hydration shells

A hydrated anion:



Mechanically, a hydration shell is a SMD with a mechanical ω_n and Q .

The lower the concentration, the larger the hydration shell around each ion: the larger its mass and the lower its ω_n .

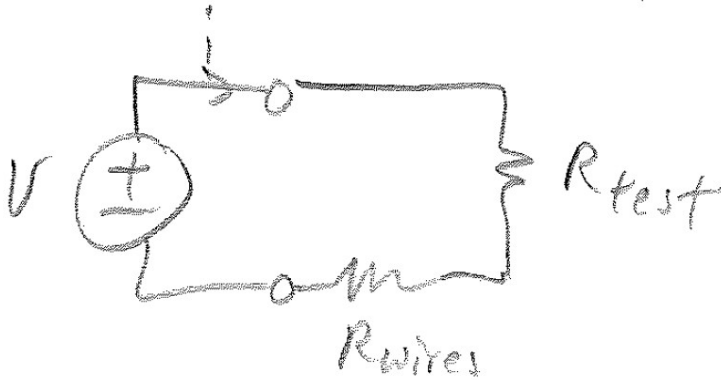
If the dipole-like hydration shell is electrically excited at its ω_n , it will vibrate. This is called relaxation and f_n is called the relaxation frequency. Electrically, the solution of hydration shells has a minimum impedance at that frequency, due to its complex permittivity.

A frequency sweep across the liquid can be performed with a network analyzer connected to an appropriate set of electrodes to detect the impedance minima. This information can sometimes be used to estimate the chemical composition and/or contaminant concentration of the aqueous solution. This is called Electrochemical Impedance Spectroscopy (EIS).

In aqueous solutions, the relaxation frequency can be quite low, in the 5 MHz to 10 MHz range.

2-Point and 4-Point Resistance Measurements

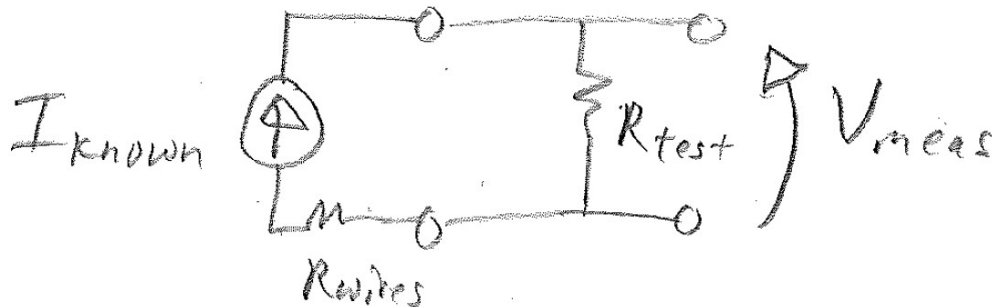
1. 2-Point Resistance Measurement



$R_{wires} \sim \text{few } \Omega$. If $R_{test} \gg R_{wires}$, then $R_{meas} = \frac{v}{i} \approx R_{test}$

However, if $R_{test} \approx R_{wires}$, then $R_{meas} = \frac{v}{i} = R_{test} + R_{wires}$, and measuring R_{meas} does not yield a good estimate for R_{test} .

2. 4-Point Resistance Measurement



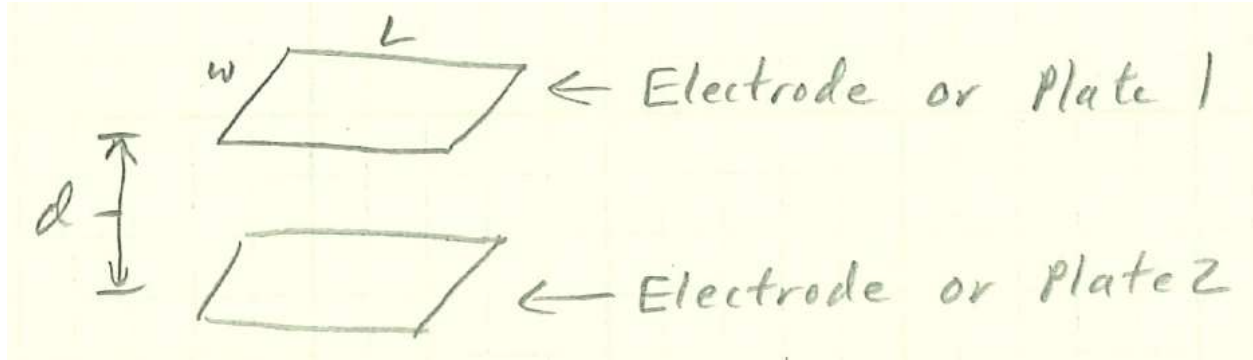
Apply a known current to the resistance under test, I_{known} . Measure the voltage across R_{test} , V_{meas} , with a high impedance voltmeter. Then compute R_{meas} to determine R_{test} :

$$R_{meas} = \frac{V_{meas}}{I_{known}} = R_{test}$$

This is a more precise resistance measurement technique, but it is also more complicated.

Capacitive Sensing

1. Simple Parallel Plate Capacitor



d is the electrode separation distance.

For $A \gg d^2$: fringing effects can be ignored and $C = \frac{\epsilon_0 \epsilon_r A}{d}$

where: ϵ_0 = permittivity of free space = 8.854 pF/m

ϵ_r = relative permittivity of the dielectric material between the electrodes

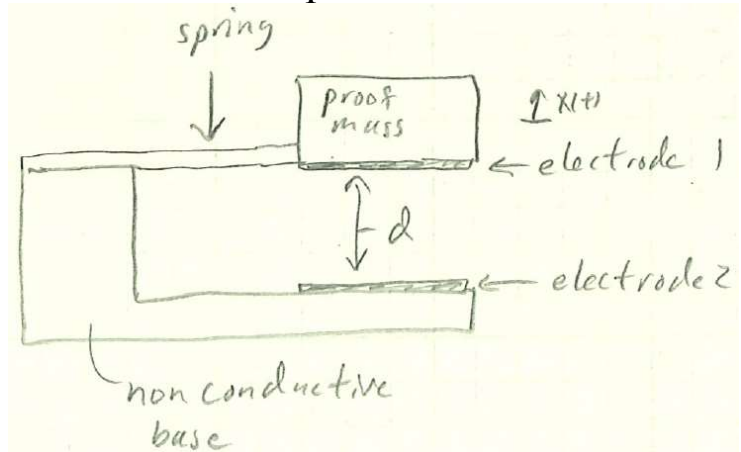
For a vacuum $\epsilon_r = 1$, and for most gases: $\epsilon_r \approx 1$

A = electrode overlapping surface area = wL here

Ways to detect with a capacitive sensor (non-fringing sensor):

- 1) Change the dielectric material, i.e. change $\epsilon_r \rightarrow$ this can be difficult to do.
- 2) Change the electrode separation distance \rightarrow easy to do.

Consider this simple MEMS accelerometer:

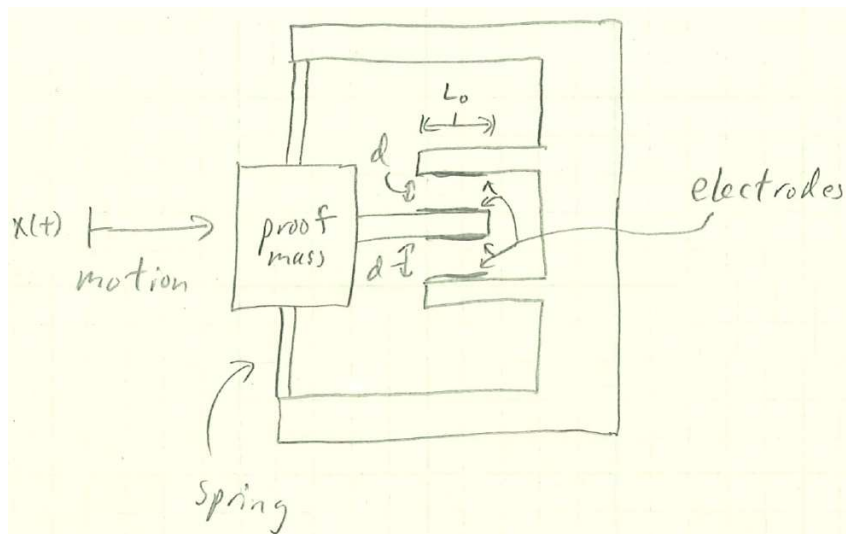


$$C = \frac{\epsilon_0 \epsilon_r A}{d_0 + x(t)}$$

where d_0 is d when $x(t) = 0$ ("at rest" condition)

3) Change the electrode overlap area \rightarrow easy to do.

Consider this MEMS accelerometer



d , w , and ϵ_r are constants here.

$$L(t) = L_0 + x(t)$$

$$\therefore C(t) = \frac{\epsilon_0 \epsilon_r W L(t)}{d} = \frac{\epsilon_0 \epsilon_r W}{d} (L_0 + x(t))$$

2. Typical Capacitor Sizes in MEMS

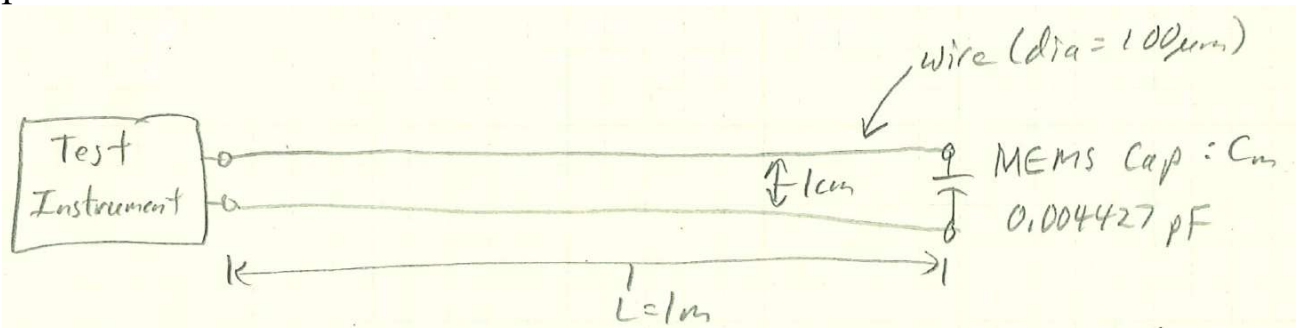
For a parallel plate capacitor: let $w = 50 \mu\text{m}$, $L = 100 \mu\text{m}$, $d = 10 \mu\text{m}$, and $\epsilon_r = 1$

$$\therefore C = \frac{(8.854)(50 \times 10^{-6})(100 \times 10^{-6})}{10 \times 10^{-6}} = 0.004427 \text{ pF} = 4.427 \text{ fF}$$

1 fF = 1×10^{-15} F ! \rightarrow a very small capacitance!

Consider that a 10% change in C is only 0.4427 fF!

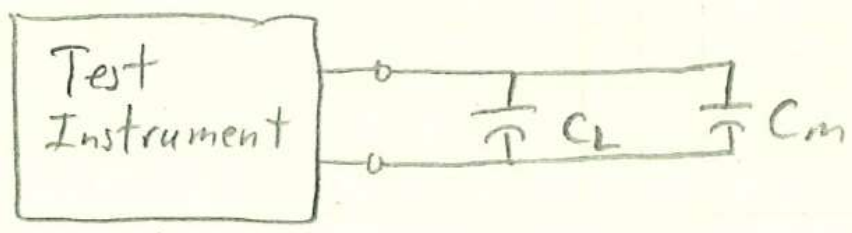
With sensor capacitances this small, stray capacitance can be a big problem. Consider this:



$$\text{For the cable: } C_L \approx \frac{(8.854)(100 \times 10^{-6})(1)}{0.01} = 0.08854 \text{ pF}$$

Observe that: $\frac{C_L}{C_m} = 20 \rightarrow C_L = 20C_m!$ or $C_L = 200\Delta C_m |_{10\% \Delta}$

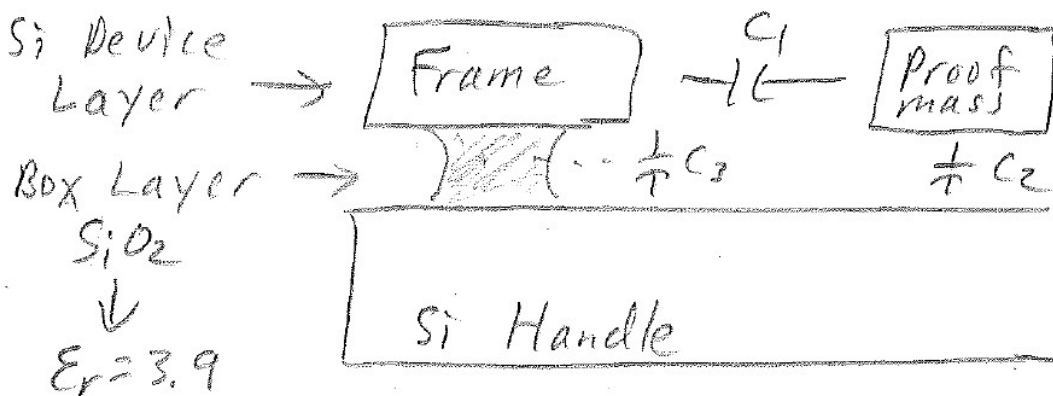
Instrumentation model:



C_m adds with C_L . Therefore, you need to locate the interface circuitry as close to the MEMS capacitor as possible, such as on the same chip.

Note: stray capacitance can also be an issue on the MEMS chip between different microstructures.

Example: A cross-section of an SOI MEMS device:



C_1 → desired capacitance between two microstructures

C_2 → stray capacitance in air ($\epsilon_r \sim 1$)

C_3 → stray capacitance in SiO_2 ($\epsilon_r = 3.9$)

So, C_2 and C_3 may not be able to be ignored, depending on the design, particularly with a typical Box Layer thickness of $0.5 \mu\text{m}$ to $1 \mu\text{m}$.

3. Capacitor Interface Circuitry: Oscillator

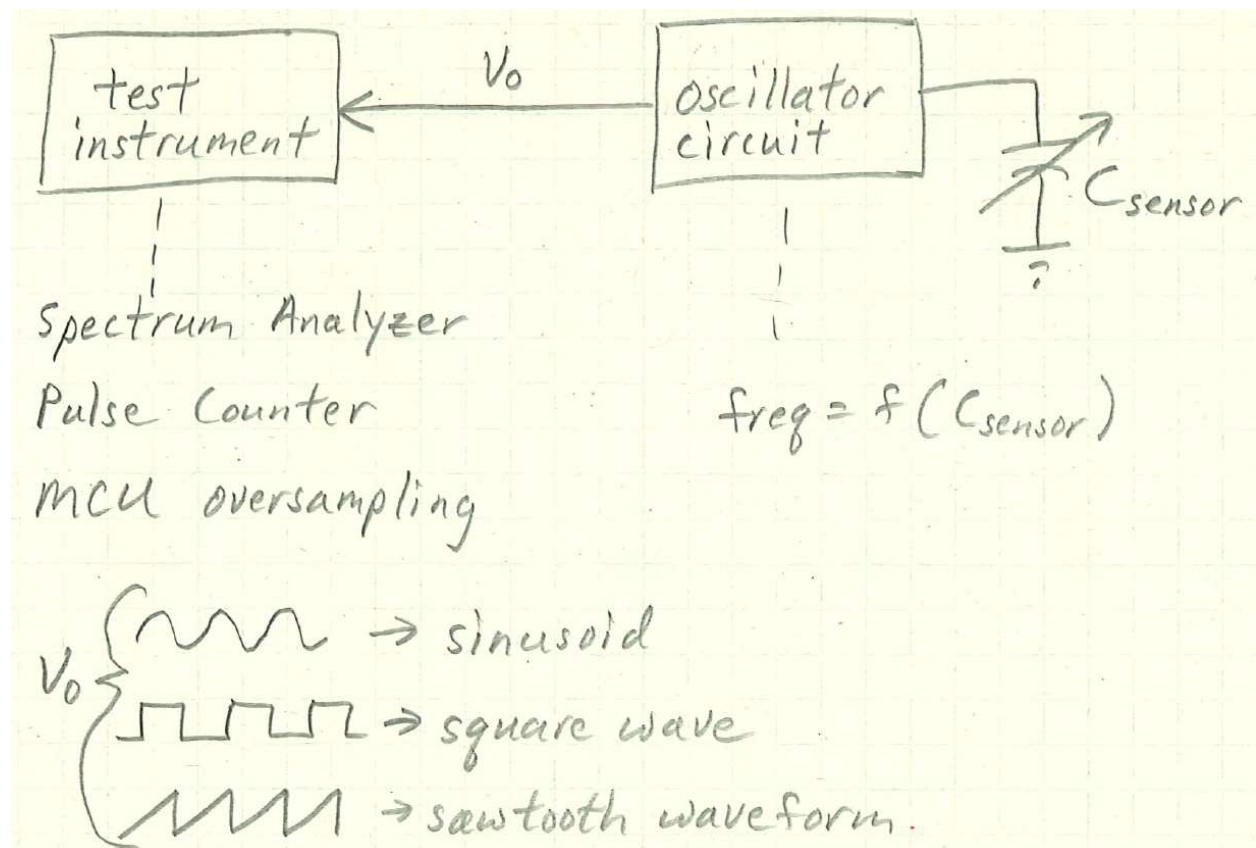
Consider C: $q = CV$ and $\dot{q} = \dot{C}V + \dot{V}C = i(t)$

In most EE application: $\dot{C} = 0 \rightarrow i(t) = C \frac{dV}{dt}$

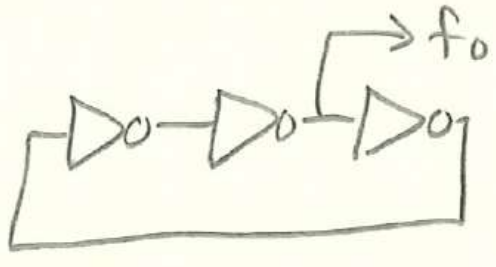
However, with capacitive sensors, $\dot{C} \neq 0$ usually

So, how then do you measure C(t)???

One technique is to place C(t) in an oscillator circuit where $\text{freq} = f(C)$:



Consider a simple CMOS ring oscillator:



It is a closed chain of an odd number of inverter stages \rightarrow unstable

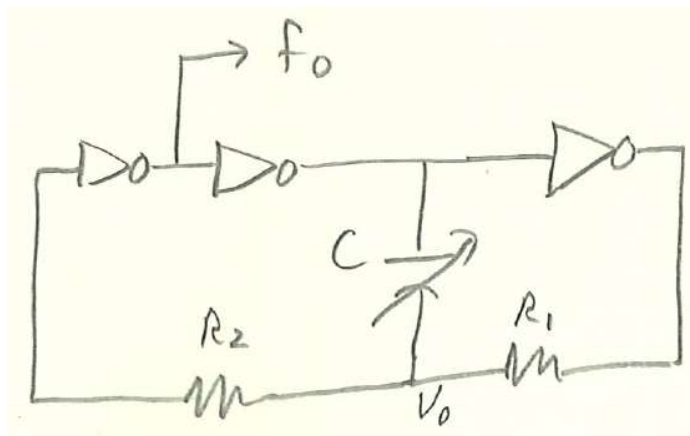
Let τ_d = propagation delay per inverter

N = number of inverters

It takes two complete cycles for two state changes (1 period)

$$\text{Therefore: } f = \frac{1}{2\tau_d N}$$

Although this oscillator circuit is not useful for a capacitive sensor interface circuit, consider this related circuit:



This oscillator circuit is called a relaxation oscillator, and it is based on an RC time constant that controls when a state change occurs in the unstable chain of CMOS inverters.

Circuit analysis:

Assume: (1) $R_1 = R_2 = R$

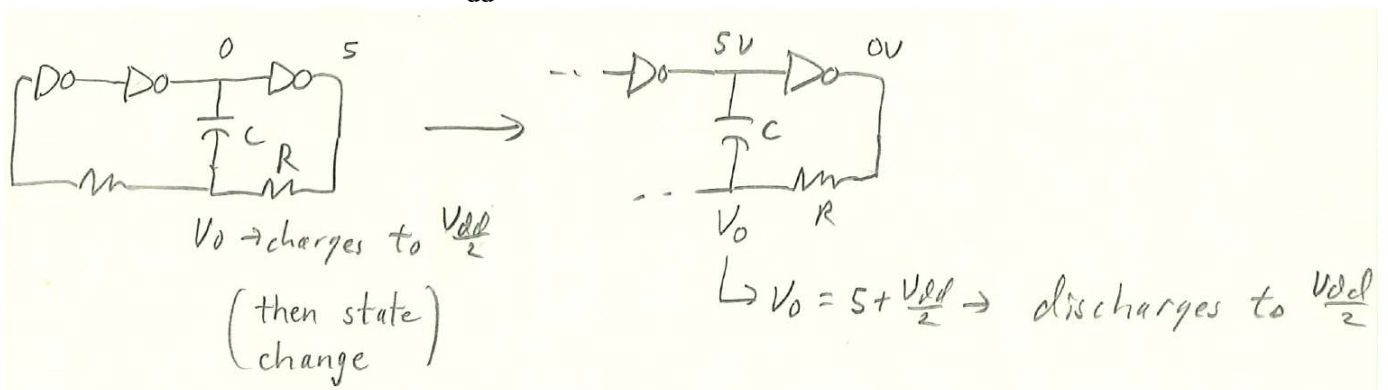
(2) $C \gg$ input capacitance on an inverter

(3) The inverter trip voltage: $V_{tr} = \frac{V_{dd}}{2}$

Therefore, when V_o crosses V_{tr} , the system changes state.

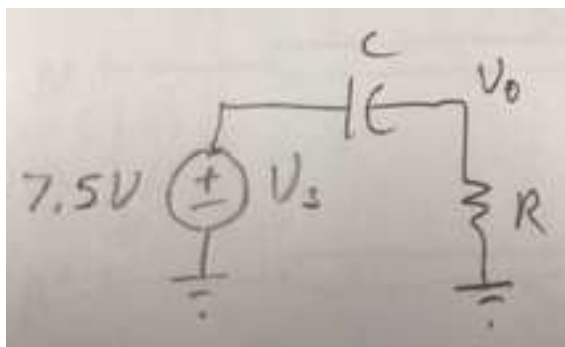
The RC time constant determines the output frequency, f_o , let's see how:

Examine the circuit with $V_{dd} = 5V$:



With $V_{dd} = 5V$: $V_{tr} = 2.5V$

At switching time, we have this circuit model:



With this model, C is fully discharged initially and $v_o = 7.5\text{V}$. We want to find t for $v_o = 2.5\text{V}$.

$$\frac{V_o}{V_s} = \frac{R}{R + \frac{1}{sC}} = \frac{s}{s + \frac{1}{RC}}$$

$$V_s(s) = \frac{7.5}{s}$$

$$V_o(s) = V_s(s) \left(\frac{s}{s + \frac{1}{RC}} \right) = \frac{7.5}{s + \frac{1}{RC}}$$

$$v_o(t) = 7.5e^{-\frac{t}{RC}}$$

We want to evaluate t when $v_o(t) = 2.5\text{ V}$

$$2.5 = 7.5e^{-\frac{t}{RC}}$$

$$\text{Solving for t: } t = -RC \ln \left(\frac{2.5}{7.5} \right) = 1.0986RC$$

For the square wave output: $T = 2t$ and $f = 1/T$, and $f = \frac{0.455}{RC}$

For an actual circuit, the equation for f will be slightly different due to the analysis assumptions. But for low frequency square waves (less than about 1 MHz), it should be close. This also works for fixed C: R-sensor.

For the case where $C = \frac{\epsilon_o \epsilon_r A}{d} = \frac{\epsilon_o \epsilon_r A}{x(t)} \rightarrow$ plate separation C sensor,

$$f = \frac{0.455}{RC} = \frac{0.455x(t)}{R\epsilon_o \epsilon_r A} \rightarrow \text{therefore } f \propto x(t)$$